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Solving problems of solution stoichiometry requires the concepts introduced in stoichiometry in Chapter 6, which also provides the basis for the discussion on reactions. (Back to the Top) 7.2 Types of Solutions. In Chapter 1, you were introduced to the concept of a mixture, which is a substance that is composed of two or more substances. Recall that mixtures can be of two types: Homogeneous ...

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Chapter 4. Stoichiometry of Chemical Reactions. 4.3 Reaction Stoichiometry. Learning Objectives. By the end of this section, you will be able to: Explain the concept of stoichiometry as it pertains to chemical reactions; Use balanced chemical equations to derive stoichiometric factors relating amounts of reactants and products ; Perform stoichiometric calculations involving mass, moles, and ...

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Stoichiometry; Types of chemical reactions; On The Exam. 7%-9% of exam score . Unit 5: Kinetics You'll explore various methods to observe the changes that occur during a chemical reaction and the effects of a series of reactions. Topics may include: Reaction rate; Introduction to rate law; Elementary reactions; Collision model; Introduction to reaction mechanisms; Multistep reaction energy ...

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